

Table 20.25 Order of introduction of element changes in evolution

Element change	Time (ago) years
$\text{H}_2\text{O}/\text{H}_2 \rightarrow \text{O}_2 (\text{O}_2^{\cdot-}, \text{H}_2\text{O}_2)$	3 to 1×10^9
$\text{CH}_4 \rightarrow \text{CO}_2$	3 to 2×10^9
$\text{WS}_4^{2-} \rightarrow \text{WO}_4^{2-}$	3×10^9
$\text{MoS}_4^{2-} \rightarrow \text{MoO}_4^{2-}$	3×10^9
$\text{H}_2\text{Se} \rightarrow \text{SeO}_4^{2-}$	3×10^9
$\text{NH}_3 \rightarrow \text{NO}(\text{NO}_3^-)$	3×10^9
$\text{Fe}^{2+} \rightarrow \text{Fe}^{3+} \downarrow, \text{H}_2\text{S} \rightarrow \text{SO}_4^{2-}$	3 to 2×10^9
$\text{ZnS} \rightarrow \text{Zn}^{2+}, \text{FeS} \rightarrow \text{Fe}_2\text{O}_3$	3 to 2×10^9
$\text{Cu}^+ \downarrow \rightarrow \text{Cu}^{2+}$	2 to 1×10^9

In niches the primitive anaerobic composition of chemicals is virtually undisturbed.