

**Table 6.3.** Energy storage density for chemical fuels. (Using data from [308])

energy storage fuel	storage density (J/m <sup>3</sup> )	storage density (J/kg)
ATP	$1.4 \times 10^8$	$1.0 \times 10^5$
H <sub>2</sub> gas, 10 <sup>3</sup> atm.	$4.9 \times 10^9$	$1.2 \times 10^8$
nitroglycerine	$1.0 \times 10^{10}$	$6.3 \times 10^6$
glycine (amino acid)	$1.0 \times 10^{10}$	$6.5 \times 10^6$
wood	$1.1 \times 10^{10}$	$1.9 \times 10^7$
urea	$1.4 \times 10^{10}$	$1.1 \times 10^7$
methanol	$1.8 \times 10^{10}$	$2.2 \times 10^7$
vegetable protein	$2.3 \times 10^{10}$	$1.7 \times 10^7$
acetone	$2.4 \times 10^{10}$	$3.1 \times 10^7$
glucose	$2.4 \times 10^{10}$	$1.6 \times 10^7$
glycogen (starch)	$2.5 \times 10^{10}$	$1.8 \times 10^7$
animal protein	$2.5 \times 10^{10}$	$1.8 \times 10^7$
carbohydrate	$2.6 \times 10^{10}$	$1.7 \times 10^7$
gasoline	$2.8 \times 10^{10}$	$4.4 \times 10^7$
butane	$3.0 \times 10^{10}$	$4.9 \times 10^7$
fat	$3.3 \times 10^{10}$	$3.9 \times 10^7$
cholesterol (lipid)	$4.2 \times 10^{10}$	$3.9 \times 10^7$
H <sub>2</sub> solid (10 <sup>5</sup> atm.)	$7.2 \times 10^{10}$	$1.2 \times 10^8$
diamond	$1.2 \times 10^{11}$	$3.3 \times 10^7$

Note that  $10^{10} \text{ J/m}^3 = 2.39 \text{ kcal/cm}^3$  and  $10^7 \text{ J/kg} = 2.39 \text{ kcal/g}$ .