

Table 4

Comparison of the entropies of biomass and Gibbs energies of combustion of biomass reported in the literature

Methods <sup>a</sup>	Statistical thermo-dynamics	Statistical mechanics	Thermo-dynamics	Low-temperature calorimetry	Roels' correlation	Battley's empirical method
Microorganisms <sup>b</sup>	Average biomass	<i>E. coli</i>	<i>E. coli</i> K-12	<i>S. cerevisiae</i>	<i>S. cerevisiae</i>	<i>S. cerevisiae</i>
Composition	CH <sub>1.8</sub> O <sub>0.5</sub> N <sub>0.2</sub>	CH <sub>1.77</sub> O <sub>0.49</sub> N <sub>0.24</sub>	CH <sub>1.59</sub> O <sub>0.374</sub> N <sub>0.263</sub> P <sub>0.023</sub> S <sub>0.006</sub>	CH <sub>1.613</sub> O <sub>0.557</sub> N <sub>0.158</sub> P <sub>0.012</sub> S <sub>0.003</sub> K <sub>0.022</sub> Mg <sub>0.003</sub> Ca <sub>0.001</sub>	CH <sub>1.613</sub> O <sub>0.557</sub> N <sub>0.158</sub> P <sub>0.012</sub> S <sub>0.003</sub> K <sub>0.022</sub> Mg <sub>0.003</sub> Ca <sub>0.001</sub>	CH <sub>1.613</sub> O <sub>0.557</sub> N <sub>0.158</sub> P <sub>0.012</sub> S <sub>0.003</sub> K <sub>0.022</sub> Mg <sub>0.003</sub> Ca <sub>0.001</sub>
$\gamma_X^0$	4.8	4.79	4.998	4.58	4.58	4.58
$S^0$ (J/K/C-mol)	112.8	13–38	94.4	34.17	48.44	34.69
$-\Delta_C G_X^0$ (kJ/C-mol) <sup>c</sup>	541.2	553–563	527.6	515.0	518.9 <sup>d</sup>	515.2
References	[34,35]	[36]	[37]	[38]	[33]	[39]

<sup>a</sup>Methods for determinations of entropies of biomass.<sup>b</sup>Microorganisms studied for estimation of the absolute entropy and the Gibbs energy of combustion.;<sup>c</sup> $\Delta_C G_X^0$  calculated according to the reported entropies of biomass.<sup>d</sup> $\Delta_C G_X^0$  calculated based on Roels' correlation.