Table 3.2 Some mid-point potentials and examples of actual redox potentials

			Oxidized $+ ne^- + mH^+ =$ reduced			
	n	m	$E_{m.7}$ (mV)	ΔE_m per pH	Typical ox/red ratio	$E_{h.7}$ (mV)[*]
Ferredoxin oxidized/reduced	1	0	−430	0		
$H^+/\frac{1}{2}H_2$ (at 1 atm)	1	1	−420	−60		
O_2 (1 atm[†])/(superoxide)	1	0	−330	0	10^{-5}	−30
$NAD^+/NADH$	2	1	−320	−30	10	−290
$NADP^+/NADPH$	2	1	−320	−30	0.01	−380
Menaquinone/menaquinol	2	2	−74	−60		
Glutathione oxidized/reduced (when GSH + GSSG = 10 mM)	2	2	−172	−60	0.01	−240[‡]
Fumarate/succinate	2	2	+30	−60		
Ubiquinone/ubiquinol	2	2	+60	−60		
Ascorbate oxidized/reduced	2	1	+60	−30		
Cyt c oxidized/reduced	1	0	+220	0		
Ferricyanide oxidized/reduced	1	0	+420	0		
O_2 (1 atm[†])/2H_2O (55 M)	4	4	+820	−60		

[*] Approximate values for mitochondrial matrix under typical conditions.
[†] 1 atm oxygen = 1.25 mM.
[‡] See equation 3.22.

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