Table 1. Reduction potentials and free energies of relevant compounds and proteins

Reduction pair	E <sub>env</sub> * (volts) <sup>a</sup>	ΔG (kJ/mol) <sup>b</sup>	Refs
Terminal electron acceptors			
Fe <sup>3+</sup> /Fe <sup>2+</sup> (pH 2)	+0.77	-74.2	[74]
Fe(III)-citrate/Fe(II)-citrate	+0.385	-37.1	[74]
Fe(III)-NTA/Fe(II)-NTA	+0.372	-35.9	[74]
Ferrihydrite <sub>solid</sub> /Fe <sup>2+</sup>	+0.1 to -0.1	-9.6 to 9.6	[74]
α-FeOOH <sub>solid</sub> /Fe <sup>2+</sup>	-0.274	26.4	[74]
α-Fe <sub>2</sub> O <sub>3solid</sub> /Fe <sup>2+</sup>	-0.287	55.4	[74]
Fe <sub>3</sub> O <sub>4 solid</sub> /Fe <sup>2+</sup>	-0.314	60.6	[74]
Fumarate/succinate	+0.033	-6.4	[75]
Electron donors relevant to Figures 1-4			
CO <sub>2</sub> /acetate	-0.29	223.8	[75]
NAD*/NADH	-0.32	61.7	[75]
Electron carriers relevant to Figure 1			
Cyc2 (pH 4.8)	+0.56	-54	[76]
Rusticyanin (pH 3.2)	+0.680	-65.6	[77]
Cyc1 (pH 3)	+0.385; +0.48	-37.1; -46.3	[78]
CycA1 (pH 4)	+0.51; +0.43	-49.2; -41.5	[72]
Components of photosynthetic electron	transport chains relevant to Figure 2		
P <sub>870</sub>	+0.45	-43.4	[16]
P <sub>870</sub> *	-1.1	106.1	[16]
Bph	-0.6	57.9	[16]
UQ <sub>A</sub>	-0.2	38.6	[16]
UQ <sub>B</sub>	+0.08	-15.4	[16]
Cytochrome bc <sub>1</sub> (b)	+0.05 and -0.09	-4.8 / 8.7	[79]
Cytochrome bc <sub>1</sub> (c <sub>1</sub> )	+0.285	-27.5	[80]
Cytochrome bc <sub>1</sub> (Rieske)	+0.28	-27	[81]
Cytochrome c <sub>2</sub>	+0.365	-35.2	[44]
Membrane components of ET chains in	Figures 1-4		
Menaquinones (ox/red)	-0.067 to -0.11	12.9 to 21.2	[82,83
Ubiquinone(ox/red)	+0.11	-21.2	[75]
Shewanella cytochromes relevant to Fig	gure 3		
CymA (pH 6)	-0.354 to -0.075	7.2 to 34.1	[47]
STC (pH 6)	-0.285 to -0.035	3.4 to 27.5	[47]
MtrA (pH 6)	-0.250 to +0.05	-4.8 to 24.1	[47]
OmcA (pH 6)	-0.325 to −0.05	4.8 to 31.4	[47]
MtrC (pH 6)	-0.275 to -0.0005	0 to 26.5	[47]
Geobacter cytochromes relevant to Figu	ure 4		
PpcA	-0.169	16.3	[84]
OmcB	-0.19	18.3	[85]
OmcS	-0.212	20.4	[52]
Endogenous and exogenous electron sh	nuttles		
Riboflavin	-0.208	40.1	[86]
Monoflavin nucleotide	-0.19	36.7	[75]
AQDS	-0.184	35.5	[87]
Humic substances	-0.2 to +0.3	-77 to 19	[87]
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 $<sup>^{</sup>a}E_{anv}^{}$ \* indicates environmentally relevant midpoint potentials: pH 7 except where noted, standard concentrations except for solid Fe minerals, for which Fe $^{2+}$  is 100  $\mu$ M.

 $<sup>^{</sup>b}\Delta G$  calculations assume standard conditions and pH 7, except in the case of iron minerals where [Fe<sup>2+</sup>] is assumed to be 100  $\mu$ M.