

Oxidation-Reduction Potentials, Absorbance Bands and Molar Absorbance of Compounds Used in Biochemical Studies

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In addition to the references (1, 2, 3, 4, 5, 7, and 8) in the table, other generally useful sources of oxidation-reduction data are: *Biochemist's Handbook*, D. Van Nostrand Co., Princeton, N.J. (1961); *The Encyclopedia of Electrochemistry*, Ed. C. A. Hampel, Reinhold Publishing Corp. N.Y. (1964); *Oxidation-Reduction Potentials in Bacteriology and Biochemistry*, L. F. Hewitt, McCorquodale and Co. Ltd., London sixth ed (1950); *Biochemisches Taschenbuch*, Springer-Verlag, N.Y., part II (1964).

The oxidation-reduction couples are listed according to decreasing values of E° or E°' . When both values are available, the order is according to E°' . Each couple is also numbered so that a particular one might be found more rapidly by consulting the index. Unless otherwise indicated, E°' is the mid-point potential for a particular couple at pH 7. Temperatures are not listed; most of the data is relevant to room temperature (20°C to 30°C). When more exact conditions are desired (ionic strength, concentration, temperature, nature of data used to derive E° or E°') the reader should consult the reference listed.

**OXIDATION-REDUCTION POTENTIALS, ABSORBANCE BANDS AND MOLAR
ABSORBANCE OF COMPOUNDS USED IN BIOCHEMICAL STUDIES**

System	E°	E°'	λ_{\max}	E_{mM}	Reference
1. $F_2(\text{gas})/F^-$	2.87	—	—	—	1
2. $H_2N_2O_2/N_2$ (gas)	2.65	—	—	—	1
3. $S_2O_8^{2-}/SO_4^{2-}$	2.0	—	—	—	1
4. H_2O_2/H_2O	1.77	—	—	—	1
5. MnO_4^-/MnO_2	1.69	—	—	—	1
6. $HClO_2/HClO$	1.64	—	—	—	1
7. H_5IO_6/IO_3^-	1.6	—	—	—	1
8. MnO_4^-/Mn^{2+}	1.51	—	—	—	1
9. Mn^{3+}/Mn^{2+}	1.4	—	—	—	2
10. $Cl_2(\text{gas})/Cl^-$	1.359	—	—	—	1
11. $ClO_2(\text{gas})/HClO_2$	1.27	—	—	—	1
12. MnO_2/Mn^{2+}	1.23	—	—	—	1
13. $[Mn^{3+}(PO_4)_2]^{3-}/[Mn^{2+}(PO_4)_2]^{4-}$	1.22	—	—	—	2
14. Pt^{2+}/Pt	1.2	—	—	—	1
15. IO_3^-/I_2	1.19	—	—	—	1
16. ClO_4^-/ClO_3^-	1.19	—	—	—	1
17. $ClO_3^-/ClO_2(\text{gas})$	1.15	—	—	—	1
18. $[Cu^{3+}(IO_6)_2]^{7-}/[Cu^{2+}(IO_6)_2]^{8-}$, pH 8 pH 12	1.1	0.7	—	—	2
19. Br_2/Br^-	1.087	—	—	—	1
20. $N_2O_4(\text{gas})/HNO_2$	1.07	—	—	—	1
21. $Fe^{3+}/Fe^{2+}O\text{-phenanthroline}$	1.06	—	—	—	3
22. $[IrCl_6]^{2-}/[IrCl_6]^{3-}$	1.05	—	—	—	1
23. VO_3^+/VO^{2+}	1.0	—	—	—	2
24. IO_4^-/IO_3^-	1.375	0.96	—	—	2
25. $HNO_2/NO(\text{gas})$	0.99	—	—	—	1
26. p-toluene-sulphochloramide, Na salt (Chloramine-T)	1.52	0.90	—	—	2
27. Nitrosoguanidine/Nitroguanidine	0.85	—	—	—	4
28. O_2/H_2O	1.229	0.816	—	—	1
29. $NO_2^-/N_2O_4(\text{gas})$	0.80	—	—	—	1
30. Ag^{+}/Ag	0.7994	—	—	—	1
31. 1,2-Benzquinone	0.792	—	—	—	5
32. Hg_2^{2+}/Hg	0.792	—	—	—	1
33. Fe^{3+}/Fe^{2+}	0.771	—	—	—	1
34. $[Mo^{3+}(CN)_6]^{3-}/[Mo^{2+}(CN)_6]^{4-}$	0.73	—	—	—	5
35. Porphyroxide	—	0.725	—	—	5
36. Pyrogallol	0.713	—	—	—	5
37. $NO(\text{gas})/H_2N_2O_2$	0.71	—	—	—	1
38. Hg^{2+}/Hg_2^{2+}	0.92	—	—	—	1
	0.709				
39. 1,4-Benzocouline	0.69938	—	—	—	5
40. 1,2-Naphthoquinone-4-sulfonate	0.628	—	—	—	5
41. Mn^{3+}/Mn^{3+} Hematoporphyrin IX, pH 9.9	—	0.626	400 (ox)	70	6
42. MnO_4^-/MnO_4^{2-}	0.6	—	—	—	1
43. $S_2O_6^{2-}/H_2SO_3$	0.6	—	—	—	1
44. $[W(CN)_8]^{3-}/[W(CN)_8]^{4-}$	0.57	—	—	—	5
45. Porphyrindin	—	0.565	—	—	5
46. NH_2OH/NH_4	—	0.562	—	—	7
47. $H_3AsO_4/HAsO_2$	0.56	—	—	—	1
48. O-tolidine	—	0.55	—	—	5
49. Cu^{2+}/Cu^+ Hemocyanin	—	0.540	350 600	—	8, 9
50. I_2/I^-	0.536	—	—	—	1
51. Cu^+/Cu	0.521	—	—	—	1
52. $S_2O_3^{2-}/S$	0.5	—	—	—	1
53. $S_2O_3^{2-}/S_2O_3^{2-}$	—	0.484	—	—	7
54. MoO_2^{2+}/MoO_3^{3+}	0.48	—	—	—	1
55. Phenylhydrazine sulfonate	0.437	—	—	—	5
56. 2-Methyl-1,4-naphthoquinone (Menadione-Vitamin K ₃)	0.422	—	—	—	5
57. P_{700}	—	0.43	—	—	10
58. $P_{890}(P_{0.44})$	—	0.44	—	—	11, 12 13, 14
59. NO_3^-/NO_2^-	0.94	0.421	—	—	1, 7
60. Cr^{3+}/Cr^{2+}	0.40	—	—	—	1

**Oxidation-Reduction Potentials, Absorbance Bands and Molar Absorbance
of Compounds Used in Biochemical Studies—(Continued)**

System	E°	E°'	λ_{\max}	E_{mM}	Reference
61. $\text{H}_2\text{SO}_3/\text{S}_2\text{O}_3^{2-}$	0.40	—	—	—	1
62. $\text{Mn}^{3+}/\text{Mn}^{2+}$ Mesoporphyrin (pyridine) ₂	—	0.387	—	—	8
63. 2,5-dihydroxy-1,4-benzoquinone	—	0.38	—	—	5
64. Adrenalin	0.809	0.380	—	—	4, 5
65. p-Amino-dimethyl aniline	—	0.38	—	—	5
66. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cytochrome f	—	0.365	413(ox) 423(red) 525(red) 555(red)	—	66
67. $[\text{Fe}(\text{CN})_6]^{3-}/[\text{Fe}(\text{CN})_6]^{4-}$	—	0.36	—	—	5
68. o-Quinone/Diphenol	—	0.35	420(ox)	1.000	15
69. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cytochrome c ₅₅₀ (<i>R. rubrum</i>)	—	0.338	409(ox) 416(red) 521(red) 550(red)	—	16
70. Cu^{2+}/Cu	0.337	—	—	—	1
71. $\text{Fe}^{3+}/\text{Fe}^{2+}$ acetate, pH 5	—	0.34	—	—	5
72. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cytochrome c _s (<i>Azotobacter</i>)	—	0.32	420(red) 526(red) 555(red)	—	17
73. $\text{As}^{5+}/\text{As}^{3+}$	—	0.316	—	—	2
74. p-Aminophenol	0.779	0.314	—	—	5
75. $\text{O}_2(\text{gas})/\text{H}_2\text{O}_2$	0.69	0.295	—	—	7
76. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cyt. c ₄ (<i>Azotobacter</i>)	—	0.30	411(ox) 416(red) 522(red) 551(red)	115.8 157.2 17.6 23.8	17
77. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cyt c ₅₅₂ (<i>Pseudomonas</i>)	—	0.300	409(ox) 416(red) 520(red) 552(red)	—	36
78. 1,4-Benzooquinone	—	0.293	—	—	5
79. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cyt a	—	0.29	—	—	18
80. p-Quinone/Hydroquinone	—	0.28	—	—	5
81. 2,6-dibromo-2'-SO ₃ H indophenol	—	0.273	—	—	5
82. $\text{Co}^{3+}/\text{Co}^{2+}$ Mesoporphyrin(pyridine) ₂	—	0.265	—	—	8
83. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Malonate, pH 4	—	0.26	—	—	5
84. 2,5-Dihydroxyphenylacetic acid (Homogentisic acid)	0.687	0.260	—	—	5
85. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Salicylate, pH 4	—	0.26	—	—	5
86. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cyt c	—	0.254	407(ox) 415(red) 521(red) 550(red)	— 125 15.9 27.7	19, 20
87. 2,6,2'-Trichloroindophenol	—	0.254	—	—	5
88. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Coproporphyrin (CN ⁻) ₂ , pH 9.6	—	0.247	—	—	8
89. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Chlorocruorin(pyridine) ₂	—	0.246	434(red) 544(red) 562(red)	—	29
90. Indophenol	—	0.228	—	—	5
91. o-Toluidine Blue	0.677	0.224	—	—	5
92. Phenol Blue	—	0.224	—	—	5
93. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cyt c ₁	—	0.22	410(ox) 418(red) 524(red) 554(red)	— 116 11.6 24.1	37
94. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Mesoporphyrin poly D,L-(lysine-phenylalanine), pH 4	—	0.22	—	—	21
95. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cyt b ₂ (yeast)	—	0.219	—	—	22
96. 2,6-Dichlorophenolindophenol (DCPIP)	—	0.217	600	20.6	5, 23
97. 2,6-Dibromoindophenol	—	0.216	—	—	5
98. Janus Green	—	0.21	—	—	5
99. 3-amino thiazine	—	0.208	—	—	5

**Oxidation-Reduction Potentials, Absorbance Bands and Molar Absorbance
of Compounds Used in Biochemical Studies—(Continued)**

System	E°	E°'	λ_{max}	E_{mM}	Reference
100. Butyryl-Co A dehydrogenase FAD ⁺ /FADH ₂ (Cu present)	—	0.187	—	—	24
101. Fe ³⁺ /Fe ²⁺ PPIX(CN ⁻) ₂ , pH 9.9	—	0.183	—	—	8
102. Fe ³⁺ /Fe ²⁺ Hemoglobin (H 6.0), pH 7.0	—	0.17	500(ox) 630(ox)	9.0 4.0	25, 26 27
103. SO ₄ ²⁻ /H ₂ SO ₃	0.17	—	—	—	1
104. 2,6-Dibromo-2'-methoxy-indophenol	—	0.161	—	—	5
105. Sn ⁴⁺ /Sn ²⁺	—	0.15	—	—	2
106. Adrenodoxin	—	0.15	414(ox)	5.7	9
107. 2,6-Dimethylindophenol	—	0.148	—	—	5
108. 1,2-Naphthoquinone	0.547	0.143	—	—	4
109. Fe ³⁺ /Fe ²⁺ PPIX(histidine) ₂ , pH 9.5	—	0.138	—	—	8
110. Fe ³⁺ /Fe ²⁺ PPIX(pyridine) ₂ ,	—	0.137	419(red) 525(red) 557(red)	192 17.5 34.4	29 30
pH 9.5	—	-0.017	—	—	8
111. 1-Naphthol-2-sulfonate indophenol	—	0.123	—	—	5
112. Fe ³⁺ /Fe ²⁺ Cyt ₅₅₃ (<i>R. sphaeroides</i>)	—	0.120	412(ox) 418(red) 523(red) 553(red)	—	31
113. Toluylene blue	—	0.115	—	—	5
114. Fe ³⁺ /Fe ²⁺ Cyt ₅₅₂ (<i>Chromatium</i>)	—	0.100	410(ox) 417(red) 525(red) 552(red)	—	16
115. Ubiquinone/Ubihydroquinone (in 95% ethanol)	0.542	0.10	275	15	32
116. TiO ²⁺ /Ti ³⁺	—	0.10	—	—	3
117. S ₄ O ₆ ²⁻ /S ₂ O ₃ ²⁻	0.08	—	—	—	1
118. Dehydroascorbic acid/ascorbic acid, pH 4	—	0.058	—	—	5
pH 8.7	—	0.166	—	—	7
119. N-methylphenazinium methosulfate (PMS)	—	-0.012	—	—	2
120. Fe ³⁺ /Fe ²⁺ Cyt b (mitochondrial)	—	0.08	387(ox) 388(semi- 450quinone)	23.8	33
—	0.077	—	—	—	34
—	0.050	429 532 561	114 21	—	—
—	—	—	—	—	18
121. [W ⁵⁺ (OH ⁻) ₄ (CN ⁻) ₄] ³⁻ / [W ⁴⁺ /OH ⁻) ₄ (CN ⁻) ₄] ⁴⁻	—	0.07	—	—	35
122. Thionine	0.563	0.064	—	—	5
123. Thioindigo-tetrasulfonate	0.409	0.063	—	—	5
124. Fe ³⁺ /Fe ²⁺ Mesoporphyrin(Pyridine) ₂	—	0.063	—	—	8
125. Phenazine ethosulphate	—	0.055	—	—	5
126. Cresyl Blue	0.583	0.047	632	—	5
127. Fe ³⁺ /Fe ²⁺ Myoglobin	—	0.046	500(ox) 630(ox)	9.1 3.5	28, 27
128. Fe ³⁺ /Fe ²⁺ Cyt b ₃ (plants)	—	0.040	560(red) 529(red)	—	39
129. 1,4-Naphthoquinone	0.470	0.036	—	—	5
130. Toluidine blue	0.534	0.034	—	—	5
131. Fumaric/Succinate	—	0.031	—	—	7
132. [Ni(C ₁₀ H ₁₀)] ⁺ /Ni(C ₁₀ H ₁₀)	—	0.03	—	—	40
133. Thiazine blue	—	0.027	—	—	5
134. Gallocyanine	—	0.021	—	—	5
135. Fe ³⁺ /Fe ²⁺ Cyt b ₅ (microsomal)	—	0.02	413(ox) 423(red) 526(red) 555(red)	117 170 13 26	41
136. Thioindigo disulfonate	0.347	0.014	—	—	5
137. Methylene blue	0.532	0.011	688(ox)	—	5

**Oxidation-Reduction Potentials, Absorbance Bands and Molar Absorbance
of Compounds Used in Biochemical Studies—(Continued)**

System	E°	E°'	λ_{max}	E_{mM}	Reference
138. $\text{Fe}^{3+}/\text{Fe}^{2+}$ methylated heme undecapeptide of Cyt c (pyridine)	—	0.008	—	—	68
139. $\text{Fe}^{3+}/\text{Fe}^{2+}$ oxalate	—	0.002	—	—	5
140. 3-Methyl-9-phenyl isoalloxazine	—	-0.002	—	—	5
141. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Hematoporphyrin (pyridine) ₂	—	-0.004	519(red) 545(red)	—	8, 29
142. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cytochromoid c (<i>Chromatium</i>)	—	-0.040	406(ox) 418(red) 525(red) 552(red)	—	42
143. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cytochromoid c (<i>R. rubrum</i>)	—	-0.008	390(ox) 424(red) 568(red)	—	70
144. Crotonyl-CoA/Butyryl-CoA	—	-0.015	—	—	38
145. Pyocyanine	0.235	-0.034 -0.038	690(ox) 370	4.5	33
146. Indigo-tetrasulfonate	0.365	-0.046	—	—	5
147. 2-Methyl-3-phytyl-1,4-naphthoquinone (Vitamin K ₁ /Dihydro-Vitamin K ₁)	0.363	-0.05	—	—	5
148. Luciferin	—	-0.05	— 490(ox) 380(ox)	— 8.85 10.8	4
149. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Rubredoxin	—	-0.057	333(ox) 311(red)	6.3 10.8	69
150. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cyt b ₆ (Chloroplasts)	—	-0.06	563(red)	—	43
151. Methyl Capri Blue	0.477	-0.061	—	—	5
152. $\text{H}_2\text{SO}_3/\text{HS}_2\text{O}_4^-$	-0.08	—	—	—	1
153. $\text{Fe}^{3+}/\text{Fe}^{2+}$ mesoporphyrin poly-D,L-(phe), pH 9	—	-0.07	—	—	21
154. Xanthine oxidase	—	-0.08	—	—	5
155. Indigo-trisulfonate	0.332	-0.081	—	—	5
156. $\text{Fe}^{3+}/\text{Fe}^{2+}$ 1, 3, 5, 8-tetramethyl porphyrin-6,7-dipropionic acid methyl ester-2,4-disulfonic acid	—	-0.09	—	—	44
157. Thiohistidine	—	-0.09	—	—	4
158. $[\text{V}(\text{C}_{10}\text{H}_{10})]^{2+}/[\text{V}(\text{C}_{10}\text{H}_{10})]^+$	—	-0.08	—	—	40
159. Glyoxylate/Glycollate	—	-0.090	—	—	7
160. $\text{Fe}^{3+}/\text{Fe}^{2+}$ heme undecapeptide from cyt c (pyridine)	—	-0.092	403(ox) 413(red) 521(red) 551(red)	117 155	68
161. 6,8,9-trimethyl isoalloxazine	—	-0.109	—	—	5
162. Chlorophyll	0.274	-0.115	—	—	5
163. $\text{CO}_2(\text{gas})/\text{CO}(\text{gas})$	-0.12	—	—	—	1
164. Yellow enzyme FMN/FMNH ₂	—	-0.122	—	—	45
165. Indigo-disulfonate	0.291	-0.125	—	—	5
166. 9-phenyl isoalloxazine	—	-0.126	—	—	4
167. Vitamin K reductase	—	-0.127	—	—	46
168. 2-OH-1,4-Naphthoquinone	—	-0.139	—	—	5
169. Thioglycolic acid	—	-0.14	—	—	4
170. $\text{Fe}^{3+}/\text{Fe}^{2+}$ (Pyrophosphate)	—	-0.14	—	—	5
171. 2-Amino-N-Methyl Phenazine Methosulfate	—	-0.145	—	—	5
172. Indigo-monosulfonate	0.262	-0.157	—	—	5
173. Hydroxypyruvate/Glycerate	—	-0.158	—	—	7
174. Oxaloacetate/Malate	—	-0.166	—	—	47
175. Brilliant Alizarin Blue	—	-0.173	—	—	5
176. Alloxazine	—	-0.170	—	—	5

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of Compounds Used in Biochemical Studies—(Continued)**

System	E°	E°'	λ_{max}	E_{mM}	Reference
177. $\text{Mn}^{3+}/\text{Mn}^{2+}$ Methyl pheophorbide <i>a</i>	—	-0.180	370(ox) 425(ox) 475(ox) 665(ox) 418(red) 647(red)	39 31 13 17 120 24	48
178. 2-Methyl-3-hydroxy-1,4-Naphthoquinone (Phthiocol)	—	-0.180	—	—	5
179. 9-Methyl-isoalloxazine	—	-0.183	—	—	4
180. Anthraquinone-2,6-disulfonate	0.228	-0.184	—	—	5
181. Pyruvate/Lactate	—	-0.185	—	—	4
		-0.190	—	—	7
182. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Protoporphyrin IX (borate buffer), pH 8.2	—	-0.188	—	—	4, 49
183. Neutral Blue	0.17	-0.19	—	—	5
184. Dihydroxy acetone-P/ α -Glycero-P	—	-0.19	—	—	4
		-0.192	—	—	7
185. Acetaldehyde/Ethanol	—	-0.197	—	—	4, 7
186. $[\text{Ti}(\text{C}_{10}\text{H}_{10})]^2+/[\text{Ti}(\text{C}_{10}\text{H}_{10})]^+$	—	-0.20	—	—	40
187. $\text{SO}_4^{2-}/\text{S}_2\text{O}_6^{2-}$	-0.2	—	—	—	1
188. $\text{Fe}^{3+}/\text{Fe}^{2+}$ heme undecapeptide, Cyt c (imidazole)	—	-0.201	—	—	68
189. Riboflavin	—	-0.208	260 375(ox) 450(ox) 419(red) 525(red) 553(red)	27.7 10.6 12.2 — 4.2	4, 50 51
190. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Cyt c ₃ (<i>Desulforibro desulfuricans</i>)	—	-0.205	410(ox) 419(red) 525(red) 553(red)	— — 140 128	52
191. $\text{Fe}^{3+}/\text{Fe}^{2+}$ heme octapeptide from Cyt c	—	-0.205	397(ox) 414(red) 520(red) 550(red)	— — 6 10	53
192. $[\text{Ru}(\text{NH}_3)_6]^{3+}/[\text{Ru}(\text{NH}_3)_6]^{2+}$	—	-0.214	—	—	68
193. $\text{Fe}^{3+}/\text{Fe}^{2+}$ heme octapeptide from Cyt c (Imidazole)	—	-0.217	405(ox) 416(red) 520(red) 550(red)	122 162	54
194. Anthraquinone-1-sulfonate	0.195	-0.218	—	—	5
195. FMN/FMNH ₂ , pH 7.09	—	-0.219	260 375(ox) 450(ox)	27.1 10.4 12.2	54 50
		-0.211	—	—	
196. FAD/FADH ₂	—	-0.219	260 375(ox) 450(ox)	37 9.3 11.3	54
		—	—	—	
197. 6,7,9-trimethyl-isoalloxazine (Lumiflavin)	—	-0.223	—	—	54
198. Janus Green B	—	-0.225	—	—	5
199. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Protoporphyrin IX (phosphate buffer), pH 8.2	—	-0.226	395(ox) 495(ox) 620(ox)	55 7 6	4, 30
200. Glutathione	—	-0.23	—	—	7, 5
		-0.34	—	—	
201. Acetoacetyl CoA/B-OH-Butyryl CoA	—	-0.238	—	—	7
202. S(rhombic)/H ₂ S	0.14	-0.243	—	—	1, 7
203. Acetyl methyl carbinol/butane-2,3-diol	—	-0.244	—	—	7
204. 3-Acetylpyridine-NAD	—	-0.248	—	—	55
205. Phenosafranine	0.280	-0.252	—	—	5
206. $\text{V}^{3+}/\text{V}^{2+}$	-0.255	—	—	—	5
207. Acetoacetate/ β -hydroxybutyrate	—	-0.346	—	—	7
208. $\text{Mn}^{3+}/\text{Mn}^{2+}$ Hematoporphyrin IX dimethyl ester	—	-0.268	—	—	6
209. $\text{Fe}^{3+}/\text{Fe}^{2+}$ Peroxidase (horseradish)	—	-0.271	415(ox) 500(ox) 640(ox)	60 10.0 3.0	56, 27, 57
210. Fructose-sorbitol	—	-0.272	—	—	5
211. $\text{H}_3\text{PO}_4/\text{H}_3\text{PO}_3$	-0.276	—	—	—	1

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of Compounds Used in Biochemical Studies—(Continued)**

System	E°	E°'	λ_{\max}	E_{mM}	Reference
212. Rosindulin 2G	0.139	-0.281	—	—	5
213. Thionicotinamide-NAD	—	-0.285	400(red)	—	58
214. Acetone/Isoproponol	—	-0.281	—	—	5
		-0.286			
215. Safranine T	0.235	-0.289	—	—	5
216. Lipoic Acid	—	-0.29	—	—	5
217. Indulin Scarlet	0.047	-0.299	—	—	5
218. Thiophenol	—	-0.30	—	—	4
219. 4-Aminoacridine	—	-0.301	—	—	59
220. Acridine	—	-0.313	—	—	59
221. NAD ⁺ /NADH	-0.105	-0.320	259(ox) 259(red) 339(red)	18 15 6.2	7, 50
			259(ox) 259(red) 339(red)	18 15 6.2	
222. NADP ⁺ /NADPH	—	-0.324	259(ox) 259(red) 339(red)	18 15 6.2	7, 50
			240(ox) (shoulder)	0.050	7
223. Neutral Red	0.240	-0.325	—	—	5
224. Cystine/Cysteine	—	-0.340	240(ox) (shoulder)	—	
225. Lipoyl dehydrogenase	—	-0.34	—	—	60
226. NAD ⁺ /α-NADH	—	-0.341	259(ox) 346(red)	17	61
227. Mn ³⁺ /Mn ²⁺ Hematoporphyrin IX	—	-0.342	370(ox) 460(ox) 545(ox) 770(ox) 416(red) 545(red)	79 50 12 1.3 175 18	6
			248.5(ox) 278 (ox)	10.2 8.9	
228. Uric acid/Xanthine	—	-0.36	—	—	7
229. Benzyl viologen	—	-0.36	—	—	5
230. Gluconolactone/Glucose	—	-0.364	—	—	7
231. 3-Aminoacridine	—	-0.369	—	—	59
232. Xanthine/Hypoxanthine	—	-0.371	248.5(ox) 278 (ox)	10.2 8.9	7, 50
			300(ox) 390(ox)	6	
233. 1-Aminoacridine	—	-0.394	—	—	59
234. Cr ³⁺ /Cr ²⁺	-0.40	—	—	—	2
235. N-methyl nicotinamide	—	-0.419	—	—	5
236. CO ₂ /Formate	-0.20	-0.42	—	—	1
237. Fe ³⁺ /Fe ²⁺ ferredoxin (<i>Clostridium</i>)	—	-0.413	300(ox) 390(ox)	6	65
238. H ⁺ /H ₂	0.000	-0.421	—	—	5
239. Fe ³⁺ /Fe ²⁺ ferredoxin (spinach)	—	-0.432	325 420(ox) 463(ox)	—	65
			—		
240. Methyl viologen	—	-0.44	—	—	5
241. Xanthine oxidase	—	-0.45	—	—	63
242. SO ₄ ²⁻ /SO ₃ ²⁻	—	-0.454	—	—	7
243. Gluconate/Glucose	—	-0.44	—	—	62
		-0.47	—		
244. 2-Aminoacridine	—	-0.486	—	—	59
245. Oxallate/Glyoxalate	—	-0.50	—	—	7
246. H ₃ PO ₃ /H ₃ PO ₂	-0.50	—	—	—	1
247. SO ₃ ²⁻ /S ₂ O ₄ ²⁻	—	-0.527	—	—	7
		-0.471	—		
248. Acetate/acetaldehyde	—	-0.581	—	—	7
		-0.589	—		
249. 2,8-diaminoacridine	—	-0.731	—	—	59
250. SiO ₂ /Si	-0.86	—	—	—	1
251. 5-Aminoacridine	—	-0.916	—	—	59
252. N ₂ (gas)/H ₃ NOH ⁺	-1.87	—	—	—	1
253. Formamidine sulfonic acid	—	-1.5	—	—	64

Oxidation-Reduction Potentials, Absorbance Bands and Molar Absorbance of Compounds Used in Biochemical Studies—(Continued)

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