TABLE III Effect of adjacent carbon groups on the rate of organic transformations a

Decarboxylation of a carboxylic acid, R-COOH or R-COO-		Dehydration of an alcohol, R-CR'R"-OH		Deacylation, isomerization and aldolization of a ketone or aldehyde, R2>C=O or R=CHO	
Substance	Estimated t _{1/2} -50 °C	Substance	Estimated t _{1/2} -50 °C	Substance	Estimated t _{1/2} -50 °C
acetic acid	3 × 10 ²⁷ s 2 × 10 ²⁵ s	1º-alcohol (ethanol, propanol, decanol, phenylethanol)	$[1\times10^{18}~\mathrm{s}]8$	2-decanone (deacyl., isomer.) decanal (0.01 M aldoliz.)	no react. (250 °C) 2 × 10 ⁶ s (250 °C)
and the same	2 ~ 10	3º-alcohol (t-butanol)	1×10^8 s	2-decanone (0.01 M aldoliz.)	4 × 10 ⁷ s (250 °C)
acetoacetic acid acetoacetate	4×10^3 s 2×10^5 s	3-hydroxy-propanal	$5\times10^5~\mathrm{s}$	acetylacetone (deacylation)	$8\times10^5~\text{s}$
lactic acid lactate	$2 \times 10^{13} \text{ s}^{\text{b}}$ $[2 \times 10^{13} \text{ s}]^{\text{c}}$	ethylene glycol	insuff, data	glyceraldehyde (isomerization) glycolaldehyde (0.01 M aldoliz.)	5×10^6 s 2×10^7 s
tartronate semialdehyde	[10 ³ s] ^d	glyceraldehyde	9×10^5 s	α-hydroxymalonaldehyde	no data found
pyruvic acid pyruvale	$1 \times 10^4 \text{ s}^e$ $6 \times 10^5 \text{ s}^e$	glycolaldehyde	aldolizes	glyoxal (intramolec. redox) pyruvaldehyde (intramolec. redox)	3×10^6 s 2×10^6 s
oxalic acid oxalate ⁻²	$6 \times 10^9 \text{ s}$ $[6 \times 10^{11} \text{ s}]^f$	lactic acid lactate	$2 \times 10^{13} \text{ s}^{\text{b}}$ $[2 \times 10^{13} \text{ s}]^{\text{c}}$	glyoxylate	decarboxylates
malonic acid malonate ⁻¹	$4 \times 10^6 \text{ s}$ $6 \times 10^6 \text{ s}$	β -hydroxy acids (β -hydroxybutyric acid, malate ⁻¹ , 2-deoxygluconic acid)	[6 × 10 ¹² s] h	malonic semialdehyde	no data found

a Unbracketed half-lifes were estimated by extrapolation using Arrhenius plots. Bracketed half-lifes were estimated by extrapolation using the Arrhenius plot slope of a related reaction

f Half-life of oxalate⁻² was estimated using the slope of the Arrhenius plot of oxalic acid.

B Half-life of 1°-alcohols was estimated from the joint Arrhenius plot of the alcohols listed.

h Half-life of β -hydroxy acids was estimated from the joint Arrhenius plot of the acids listed.